

- 4.46** (a) Silicon at $T = 300$ K is uniformly doped with boron atoms to a concentration of $3 \times 10^{16} \text{ cm}^{-3}$ and with arsenic atoms to a concentration of $1.5 \times 10^{16} \text{ cm}^{-3}$. Is the material n type or p type? Calculate the thermal equilibrium concentrations of majority and minority carriers. (b) Additional impurity atoms are added such that holes are the majority carrier and the thermal equilibrium concentration is $p_0 = 5 \times 10^{16} \text{ cm}^{-3}$. What type and concentration of impurity atoms must be added? What is the new value of n_0 ?

a) From Table 4.3, boron is an acceptor while arsenic is a donor.

$$N_a = 3 \times 10^{16} \text{ \# / cm}^3 > N_d = 1.5 \times 10^{16} \text{ \# / cm}^3 \Rightarrow \underline{\underline{\text{p-type}}}$$

Per Table B.4, $n_i = 1.5 \times 10^{10} \text{ \# / cm}^3$ @ 300K

$$\begin{aligned} (4.62) \quad p_0 &= \frac{N_a - N_d}{2} + \sqrt{\left(\frac{N_a - N_d}{2}\right)^2 + n_i^2} \\ &= \frac{(3 - 1.5) \times 10^{16}}{2} + \sqrt{\left(\frac{1.5 \times 10^{16}}{2}\right)^2 + (1.5 \times 10^{10})^2} \\ \underline{\underline{p_0}} &= \underline{\underline{1.5 \times 10^{16} \text{ \# / cm}^3}} = \underline{\underline{1.5 \times 10^{22} \text{ \# / m}^3}} \end{aligned}$$

$$(4.43) \quad n_0 = \frac{n_i^2}{p_0} = \frac{(1.5 \times 10^{10})^2}{1.5 \times 10^{22}}$$

$$\underline{\underline{n_0 = 1.5 \times 10^4 \text{ \# / cm}^3}} = \underline{\underline{1.5 \times 10^{10} \text{ \# / m}^3}}$$

b) In (4.62), we saw that n_i is negligible. To achieve $p_0 = 5 \times 10^{16} \text{ \# / cm}^3 > 1.5 \times 10^{16} \text{ \# / cm}^3$, we must add more acceptors (boron).

$$p_0 = 5 \times 10^{16} = \Delta N_a + 3 \times 10^{16} - 1.5 \times 10^{16}$$

$$\hookrightarrow \underline{\underline{\Delta N_a = 3.5 \times 10^{16} \text{ \# / cm}^3}} = \underline{\underline{3.5 \times 10^{22} \text{ \# / m}^3}}$$

$$n_0 = \frac{(1.5 \times 10^{10})^2}{5 \times 10^{16}} \Rightarrow \underline{\underline{n_0 = 4.5 \times 10^3 \text{ \# / cm}^3}} = \underline{\underline{4.5 \times 10^9 \text{ \# / m}^3}}$$