

- 4.20** (a) If  $E_c - E_f = 0.28$  eV in gallium arsenide at  $T = 375$  K, calculate the values of  $n_0$  and  $p_0$ . (b) Assuming the value of  $n_0$  in part (a) remains constant, determine  $E_c - E_f$  and  $p_0$  at  $T = 300$  K.

➤ Assume  $E_g$  is temperature independent.

- a) From Table B.4, we get  $N_c = 4.7 \times 10^{17}$  #/cm<sup>3</sup>,  $N_v = 7.0 \times 10^{18}$  #/cm<sup>3</sup>, and  $E_g = 1.42$  eV at 300K for GaAs.

Per (4.10)  $N_c = 2 \left( \frac{2\pi m_n^* k_B T}{h^2} \right)^{3/2}$ , we can scale  $N_c$  to 375 K as

$$N_c = 4.7 \times 10^{17} \left( \frac{375}{300} \right)^{3/2} = 5.59017 \times 10^{17} \text{ #/cm}^3.$$

Per (4.11),  $n_0 = N_c e^{-(E_c - E_f)/k_B T} = 5.59017 \times 10^{17} e^{-(0.28)/[(8.617333 \times 10^{-5} \text{ eV/K}) 375 \text{ K}]}$

$$\Rightarrow \underline{n_0 = 9.64696 \times 10^{13} \text{ #/cm}^3}.$$

Per (4.18)  $N_v = 2 \left( \frac{2\pi m_p^* k_B T}{h^2} \right)^{3/2}$ , we can scale  $N_v$  to 375 K as

$$N_v = 7.0 \times 10^{18} \left( \frac{375}{300} \right)^{3/2} = 9.782797 \times 10^{18} \text{ #/cm}^3.$$

Per (4.19),  $p_0 = N_v e^{-(E_f - E_v)/k_B T}$  where  $E_f - E_v = E_g - (E_c - E_f) = 1.42 - 0.28$  eV.

$$p_0 = N_v e^{-(E_f - E_v)/k_B T} = 9.782797 \times 10^{18} e^{-(1.42 - 0.28)/[(8.617333 \times 10^{-5} \text{ eV/K}) 375 \text{ K}]}$$

$$\Rightarrow \underline{p_0 = 4672.367 \text{ #/cm}^3}.$$

- b) At 300 K, assuming  $n_0 = 9.64696 \times 10^{13}$  #/cm<sup>3</sup>, we can solve (4.11) to get

$$E_c - E_f = -k_B T \ln \left( \frac{n_0}{N_c} \right) = k_B T \ln \left( \frac{N_c}{n_0} \right)$$

$$= (8.617333 \times 10^{-5} \text{ eV/K}) 300 \text{ K} \ln \left( \frac{4.7 \times 10^{17}}{9.64696 \times 10^{13}} \right)$$

$$\Rightarrow \underline{E_c - E_f = 0.2195166 \text{ eV}}.$$

Per (4.19),  $p_0 = N_v e^{-(E_f - E_v)/k_B T}$  where  $E_f - E_v = E_g - (E_c - E_f) = 1.42 - 0.2195166$  eV.

$$p_0 = N_v e^{-(E_f - E_v)/k_B T} = 7.0 \times 10^{18} e^{-(1.42 - 0.2195166)/[(8.617333 \times 10^{-5} \text{ eV/K}) 300 \text{ K}]}$$

$$\Rightarrow \underline{p_0 = 0.0476272 \text{ #/cm}^3}.$$