

3.27 (a) Determine the total number (#/cm³) of energy states in silicon between E_v and $E_v - 3kT$ at (i) $T = 300$ K and (ii) $T = 400$ K.

➤ Repeat part a) for Germanium.

c) Per (3.75), $g_v(E) = \frac{4\pi(2m_p^*)^{3/2}}{h^3} \sqrt{E_v - E}$. Then, following a process similar to Example 3.4, the number of valence energy states in Germanium between E_v and $E_v - 3k_B T$ is

$$\begin{aligned} N_V &= \int_{E_v - 3k_B T}^{E_v} g_v(E) dE = \int_{E_v - 3k_B T}^{E_v} \frac{4\pi(2m_p^*)^{3/2}}{h^3} \sqrt{E_v - E} dE \\ &= \left[\frac{4\pi(2m_p^*)^{3/2}}{h^3} \frac{-2}{3} (E_v - E)^{3/2} \right]_{E_v - 3k_B T}^{E_v} = \frac{-8\pi(2m_p^*)^{3/2}}{3h^3} \left[(E_v - E_v)^{3/2} - (E_v - E_v + 3k_B T)^{3/2} \right] \\ &= \frac{8\pi(2m_p^*)^{3/2}}{3h^3} (3k_B T)^{3/2} \end{aligned}$$

Per Table B.4, the density of states effective hole mass for Germanium is $m_p^* = 0.37m_0$. Using this value and other constants, we get

$$\begin{aligned} N_v &= \frac{8\pi \left(2(0.37)9.1093837 \times 10^{-31} \right)^{3/2}}{3 \left(6.62607015 \times 10^{-34} \right)^3} \left[3(1.380649 \times 10^{-23}) \right]^{3/2} T^{3/2} \\ &= 4.248527 \times 10^{21} T^{3/2} \end{aligned}$$

(i) At $T = 300$ K, $N_v = 4.248527 \times 10^{21} (300)^{3/2}$

$$\Rightarrow \underline{N_v = 2.2076 \times 10^{25} \text{ \#/m}^3 = 2.2076 \times 10^{19} \text{ \#/cm}^3}.$$

(ii) At $T = 400$ K, $N_v = 4.248527 \times 10^{21} (400)^{3/2}$

$$\Rightarrow \underline{N_v = 3.39882 \times 10^{25} \text{ \#/m}^3 = 3.39882 \times 10^{19} \text{ \#/cm}^3}.$$