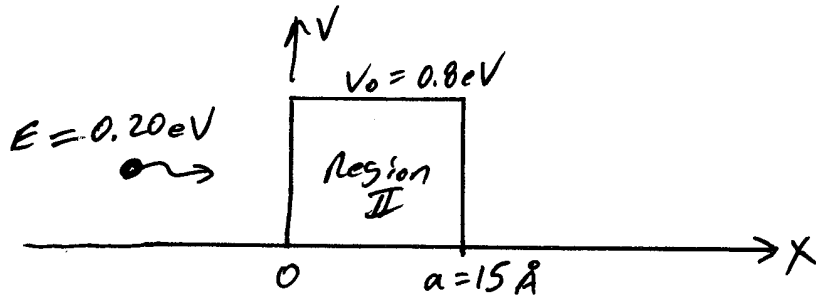


- 2.36 (a) Estimate the tunneling probability of a particle with an effective mass of  $0.067m_0$  (an electron in gallium arsenide), where  $m_0$  is the mass of an electron, tunneling through a rectangular potential barrier of height  $V_0 = 0.8$  eV and width  $15 \text{ \AA}$ . The particle kinetic energy is  $0.20$  eV. (b) Repeat part (a) if the effective mass of the particle is  $1.08m_0$  (an electron in silicon).



From notes,  $T = \frac{1}{1 + \frac{V_0^2 \sinh^2(\kappa_2 a)}{4E(V_0 - E)}} \quad (\text{exact})$

(2.63)  $T \approx 16 \left(\frac{E}{V_0}\right) \left(1 - \frac{E}{V_0}\right) e^{-2\kappa_2 a} \quad (\text{approx})$

where (2.61b)  $\kappa_2 = \sqrt{\frac{2m(V_0 - E)}{\hbar^2}}$

a)  $m = 0.067m_0$

$$\kappa_2 = \sqrt{\frac{2(0.067)(9.1093837 \times 10^{-31})(0.8 - 0.2)(1.6021766 \times 10^{-19})}{(1.054571817 \times 10^{-34})^2}}$$

$$= 1.027192 \times 10^9 \text{ Np/m}$$

$$\kappa_2 a = 1.0272 \times 10^9 (15 \times 10^{-10}) = 1.54079 \quad \leftarrow \text{approx. eqn not going to be great}$$

$$(\text{exact}) T_a = \frac{1}{1 + \frac{0.8^2 \sinh^2(1.54079)}{4(0.2)(0.8 - 0.2)}} \Rightarrow \underline{\underline{T_a = 0.131357}}$$

$$(\text{approx}) T_a \approx 16 \left(\frac{0.2}{0.8}\right) \left(1 - \frac{0.2}{0.8}\right) e^{-2(1.54079)} \Rightarrow \underline{\underline{T_a \approx 0.137661}}$$

$\Rightarrow$  Fair agreement

$$b) m = 1.08 m_0$$

$$\text{Now } \kappa_2 = \sqrt{\frac{2(1.08)9.1093837 \times 10^{-31}(0.8-0.2)1.6021766 \times 10^{-19}}{(1.054571817 \times 10^{-34})^2}}$$

$$= 4.12407 \times 10^9 \text{ N/m}$$

$$\kappa_2 a = 4.124 \times 10^9 (15 \times 10^{-10}) = 6.18611 \text{ } \approx \text{expect approx. to be good}$$

$$\text{(exact) } T_b = \frac{1}{1 + \frac{0.8^2 \sinh^2(6.18611)}{4(0.2)(0.8-0.2)}} \Rightarrow \underline{\underline{T_b = 1.270389 \times 10^{-5}}}$$

$$\text{(approx) } T_b \approx 16 \left(\frac{0.2}{0.8}\right) \left(1 - \frac{0.2}{0.8}\right) e^{-2(6.18611)} \Rightarrow \underline{\underline{T_b \approx 1.270395 \times 10^{-5}}}$$

$\Rightarrow$  Excellent agreement