

Example- Let's examine the various quantities related to putting an electron in an infinite potential well that is 4 Angstroms wide.

Define some constants

$$h := 6.62607015 \cdot 10^{-34} \text{ J-s} \quad h_{\text{mod}} := \frac{h}{2 \cdot \pi} \quad h_{\text{mod}} = 1.05457 \times 10^{-34} \text{ J-s}$$

$$m := 9.1093837015 \cdot 10^{-31} \text{ kg} \quad a := 4 \cdot 10^{-10} \text{ m}$$

Calculate some quantities and/or define equations

Per (2.33), the wave number (rad/m), in terms of quantum number n , is $k(n) := \frac{n \cdot \pi}{a}$

Per (2.35) $A_2 := \sqrt{\frac{2}{a}}$ $A_2 = 7.071 \times 10^4$ $\text{m}^{-0.5}$

Per (2.38), the quantized energy (J & eV), in terms of quantum number n , is

$$E(n) := \frac{h_{\text{mod}}^2 \cdot n^2 \cdot \pi^2}{2 \cdot m \cdot a^2} \quad E_{\text{eV}}(n) := \frac{E(n)}{1.602176634 \cdot 10^{-19}}$$

Per (2.39), the wave function ($\text{m}^{-0.5}$), in terms of quantum number n and position x , is

$$\psi(n, x) := A_2 \cdot \sin(k(n) \cdot x)$$

Per (2.17), the probability density function (m^{-1}), in terms of quantum number n and position x , is

$$\Psi^2(n, x) := \overline{\psi(n, x) \cdot \psi(n, x)}$$

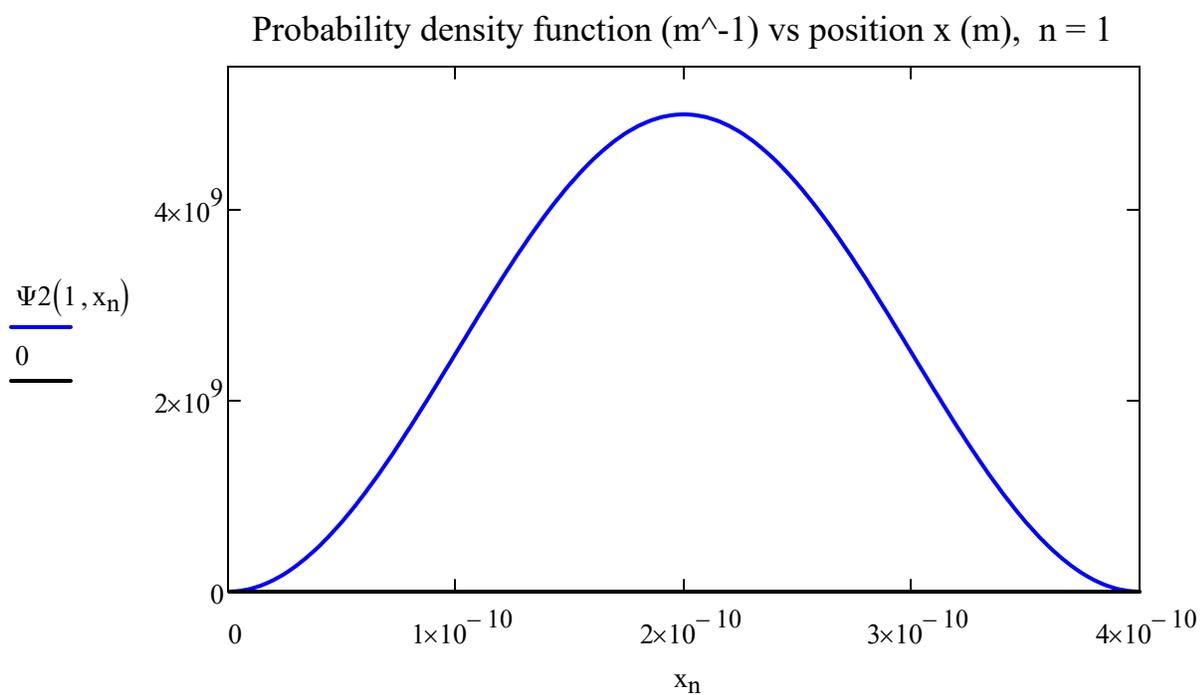
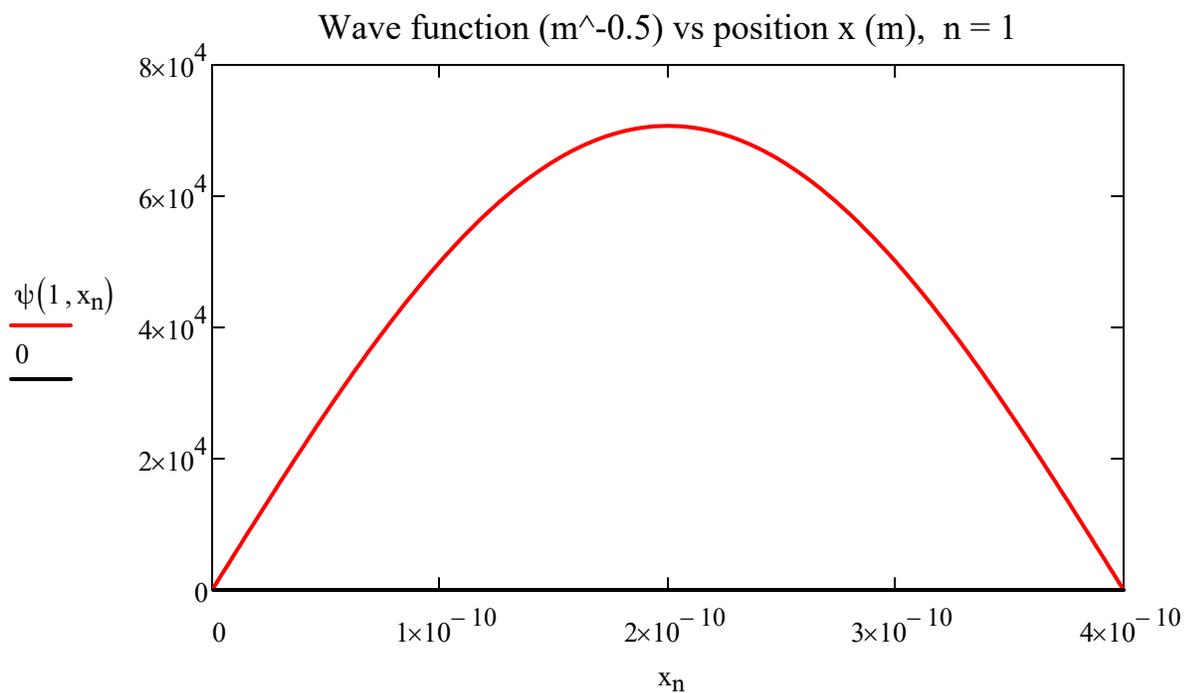
Define a vector of positions x across the width of the potential well, i.e., $0 \leq x \leq a$.

$$N_{\text{max}} := 200 \quad n := 0..N_{\text{max}} \quad x_n := \frac{n}{N_{\text{max}}} \cdot a$$

For the first quantum level, i.e., $n = 1$

$$k(1) = 7.854 \times 10^9 \quad (\text{rad/m})$$

$$E(1) = 3.765 \times 10^{-19} \quad (\text{J}) \quad \text{or} \quad E_eV(1) = 2.3502 \quad (\text{eV})$$



For the second quantum level, i.e., $n = 2$

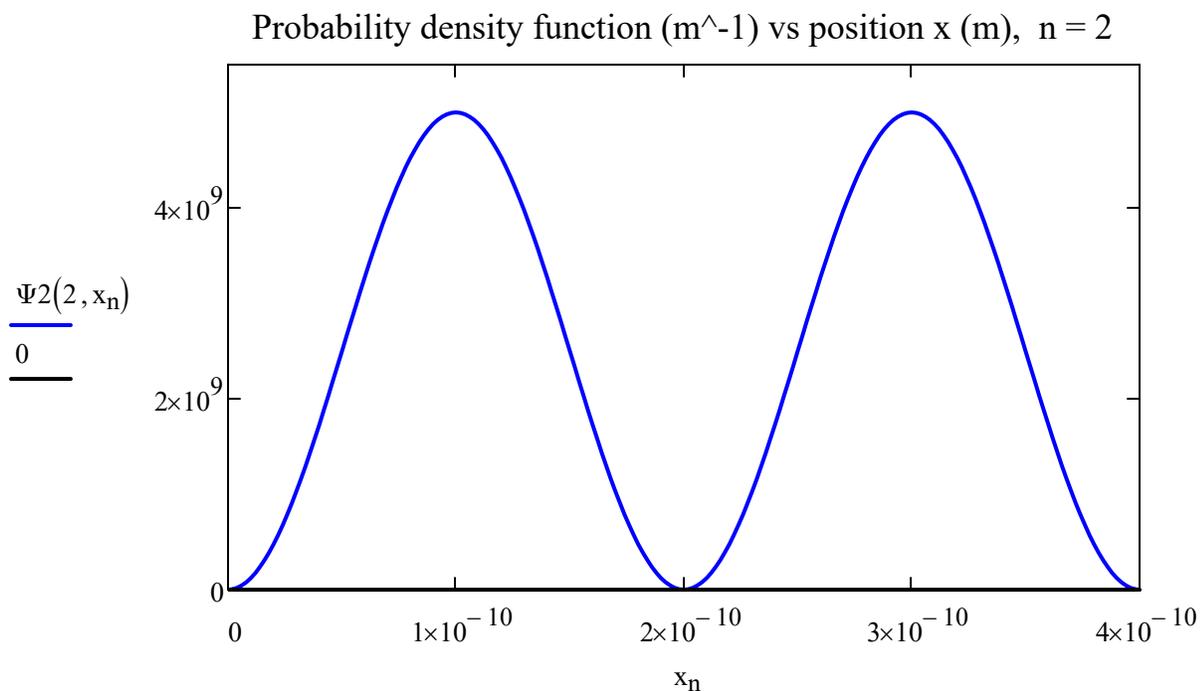
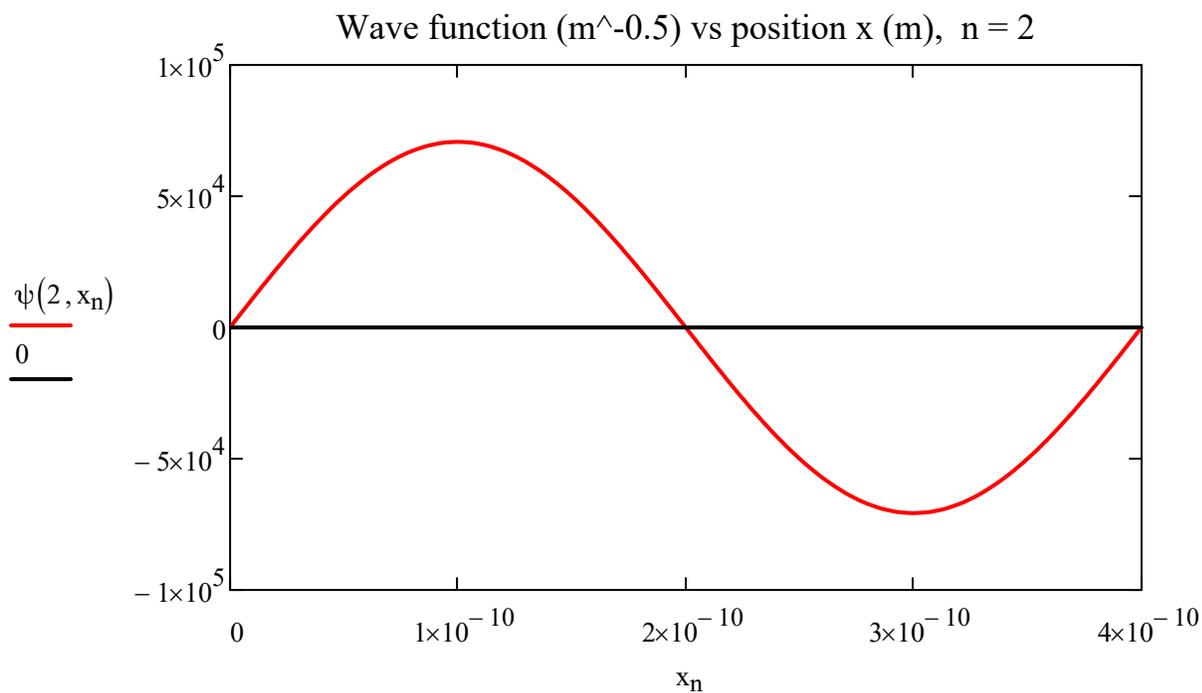
$$k(2) = 1.5708 \times 10^{10} \quad (\text{rad/m})$$

$$E(2) = 1.506 \times 10^{-18} \quad (\text{J})$$

or

$$EeV(2) = 9.4008 \quad (\text{eV})$$

$$\frac{EeV(2)}{EeV(1)} = 4$$



For the third quantum level, i.e., $n = 3$

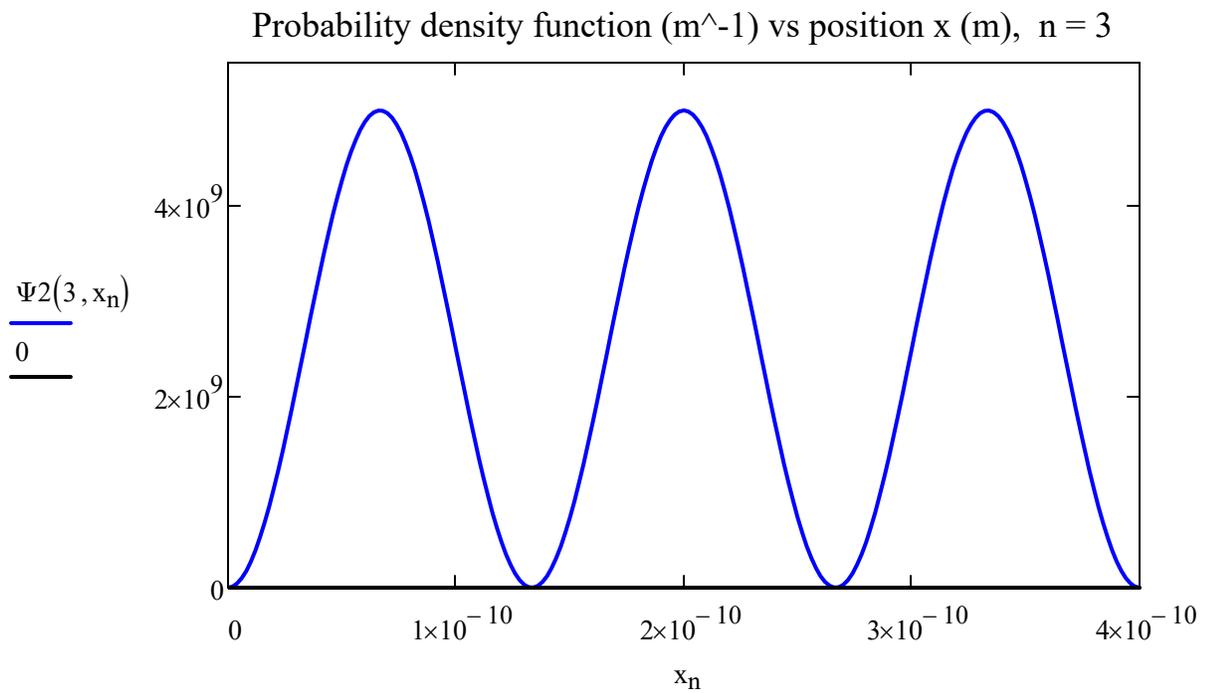
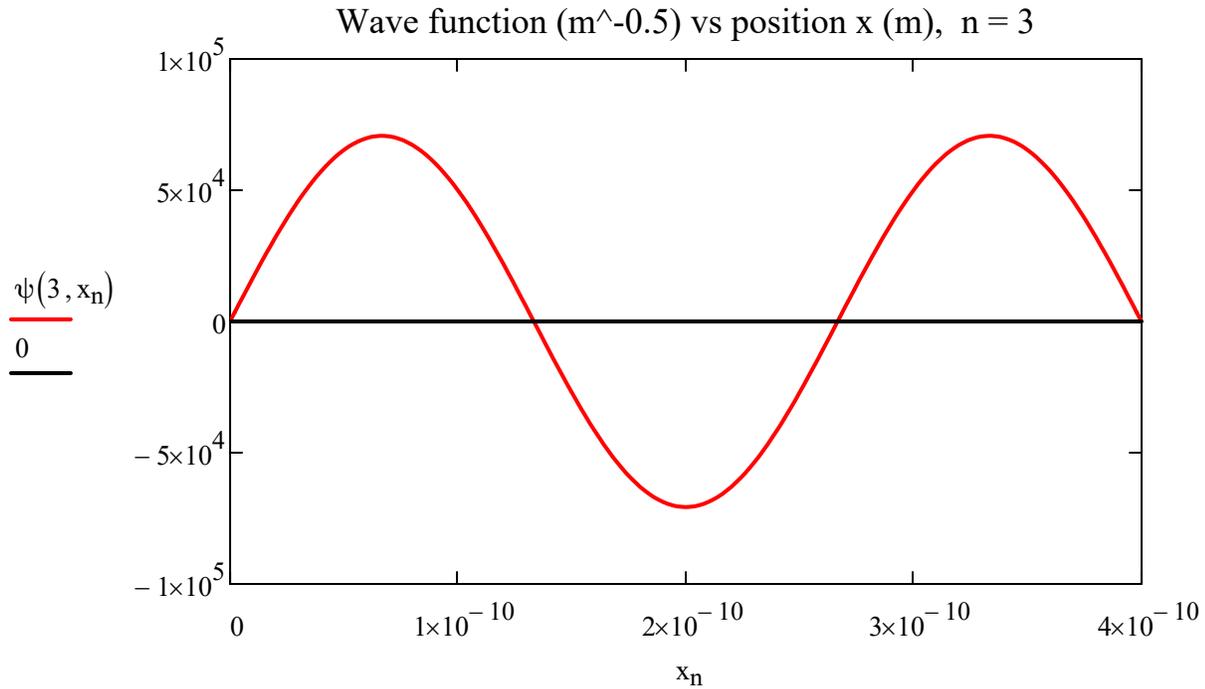
$$k(3) = 2.3562 \times 10^{10} \text{ (rad/m)}$$

$$E(3) = 3.389 \times 10^{-18} \text{ (J)}$$

or

$$EeV(3) = 21.1517 \text{ (eV)}$$

$$\frac{EeV(3)}{EeV(1)} = 9$$



For the fourth quantum level, i.e., $n = 4$

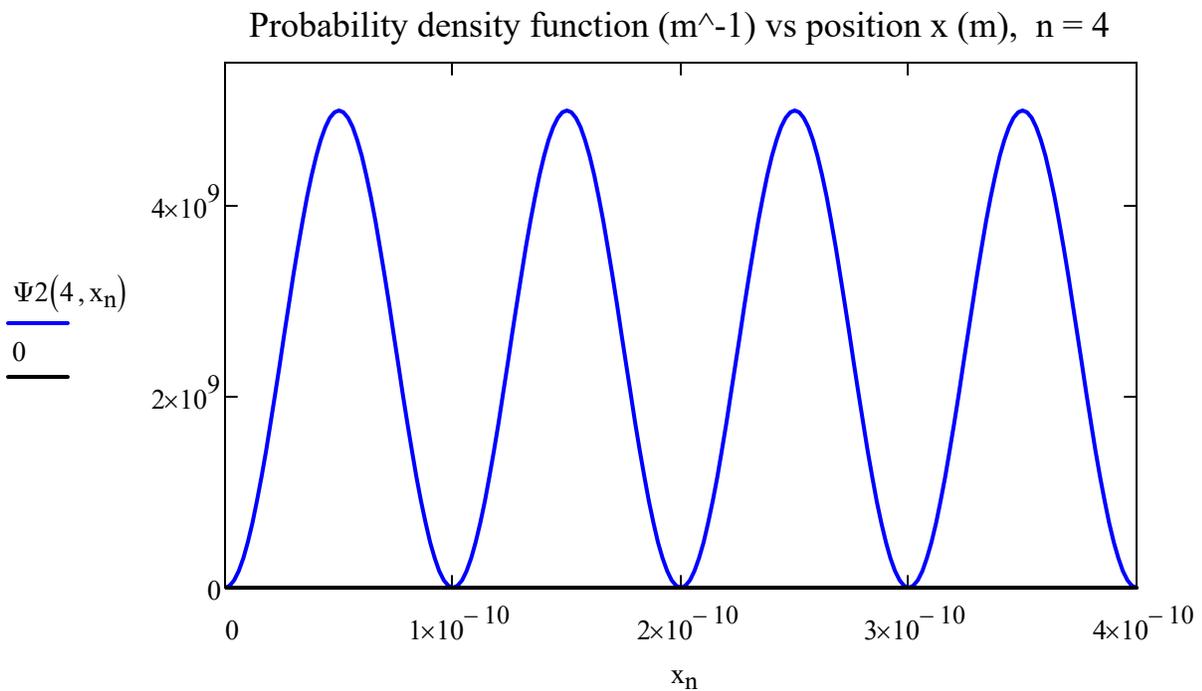
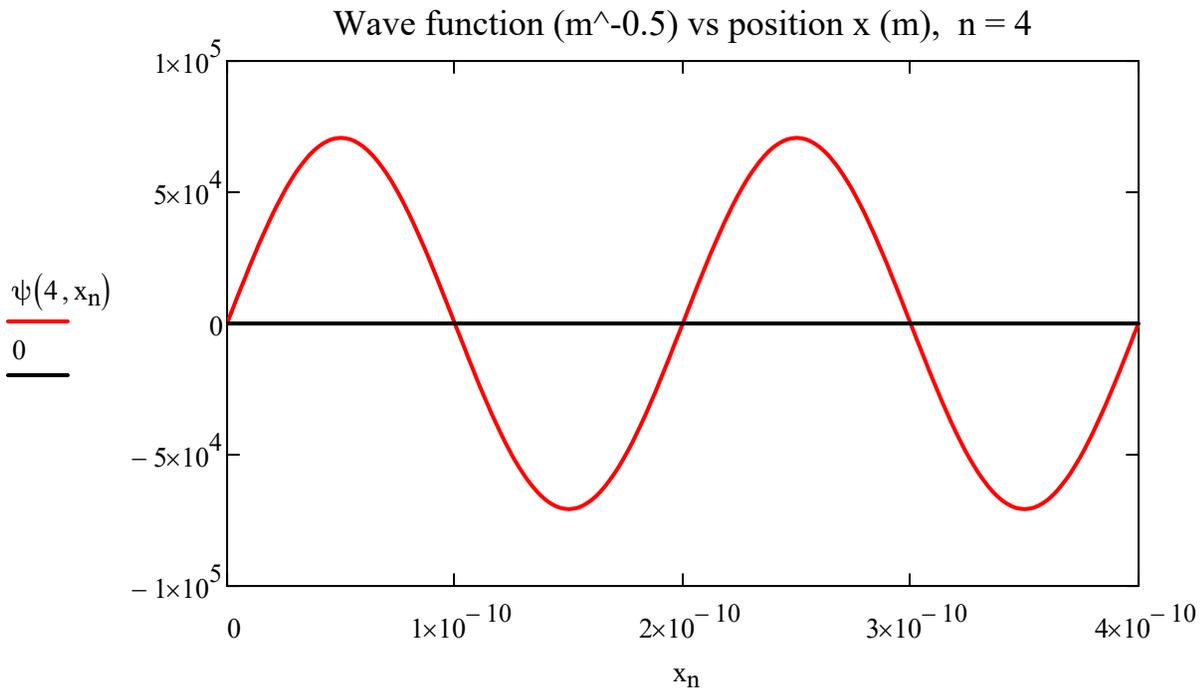
$$k(4) = 3.1416 \times 10^{10} \quad (\text{rad/m})$$

$$E(4) = 6.025 \times 10^{-18} \quad (\text{J})$$

or

$$EeV(4) = 37.603 \quad (\text{eV})$$

$$\frac{EeV(4)}{EeV(1)} = 16$$



For the fifth quantum level, i.e., $n = 5$

$$k(5) = 3.927 \times 10^{10} \quad (\text{rad/m})$$

$$E(5) = 9.414 \times 10^{-18} \quad (\text{J})$$

or

$$EeV(5) = 58.7547 \quad (\text{eV})$$

$$\frac{EeV(5)}{EeV(1)} = 25$$

