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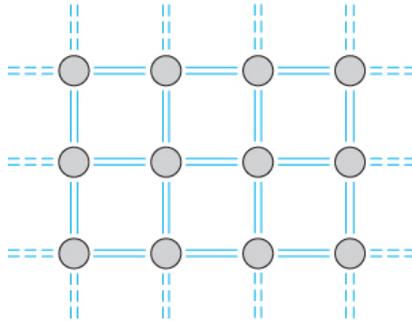
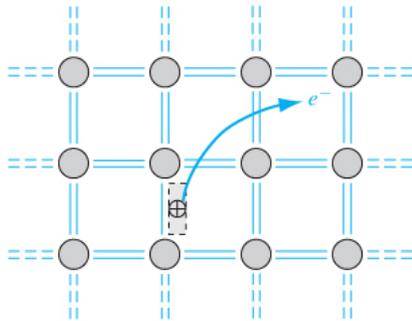
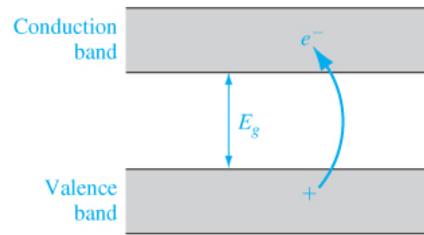


Figure 3.12 | Two-dimensional representation of the covalent bonding in a semiconductor at $T = 0$ K.

➤ Now, let temperature increase so that $K.E. = T > 0$



(a)



(b)

Figure 3.13 | (a) Two-dimensional representation of the breaking of a covalent bond. (b) Corresponding line representation of the energy band and the generation of a negative and positive charge with the breaking of a covalent bond.

➤ For silicon, we might see

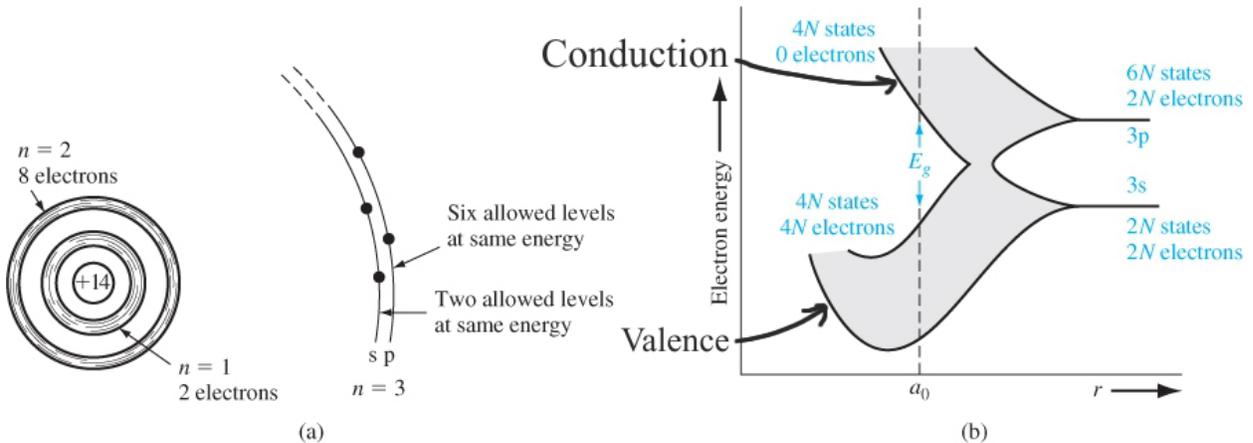


Figure 3.4 | (a) Schematic of an isolated silicon atom. (b) The splitting of the 3s and 3p states of silicon into the allowed and forbidden energy bands. (From Shockley [6].)

- Looking at energy E versus k bands at a) 0 K and b) > 0 K, we see that some of the valence electrons in the highest energy states move to some of the lowest energy conduction states.
- Note the distribution is shown as symmetric as we are assuming there is no externally applied forces or fields.

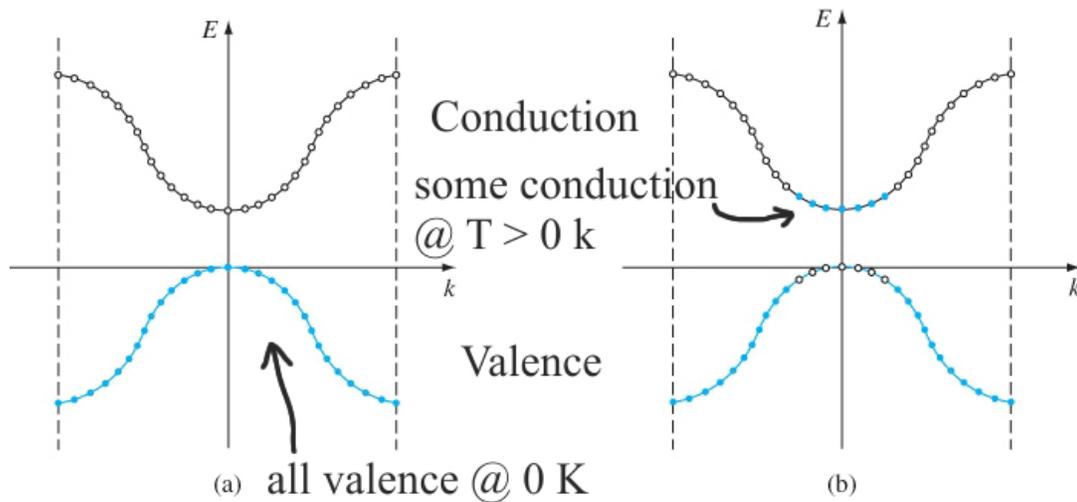


Figure 3.14 | The E versus k diagram of the conduction and valence bands of a semiconductor at (a) $T = 0$ K and (b) $T > 0$ K.